#### Ventilation Management

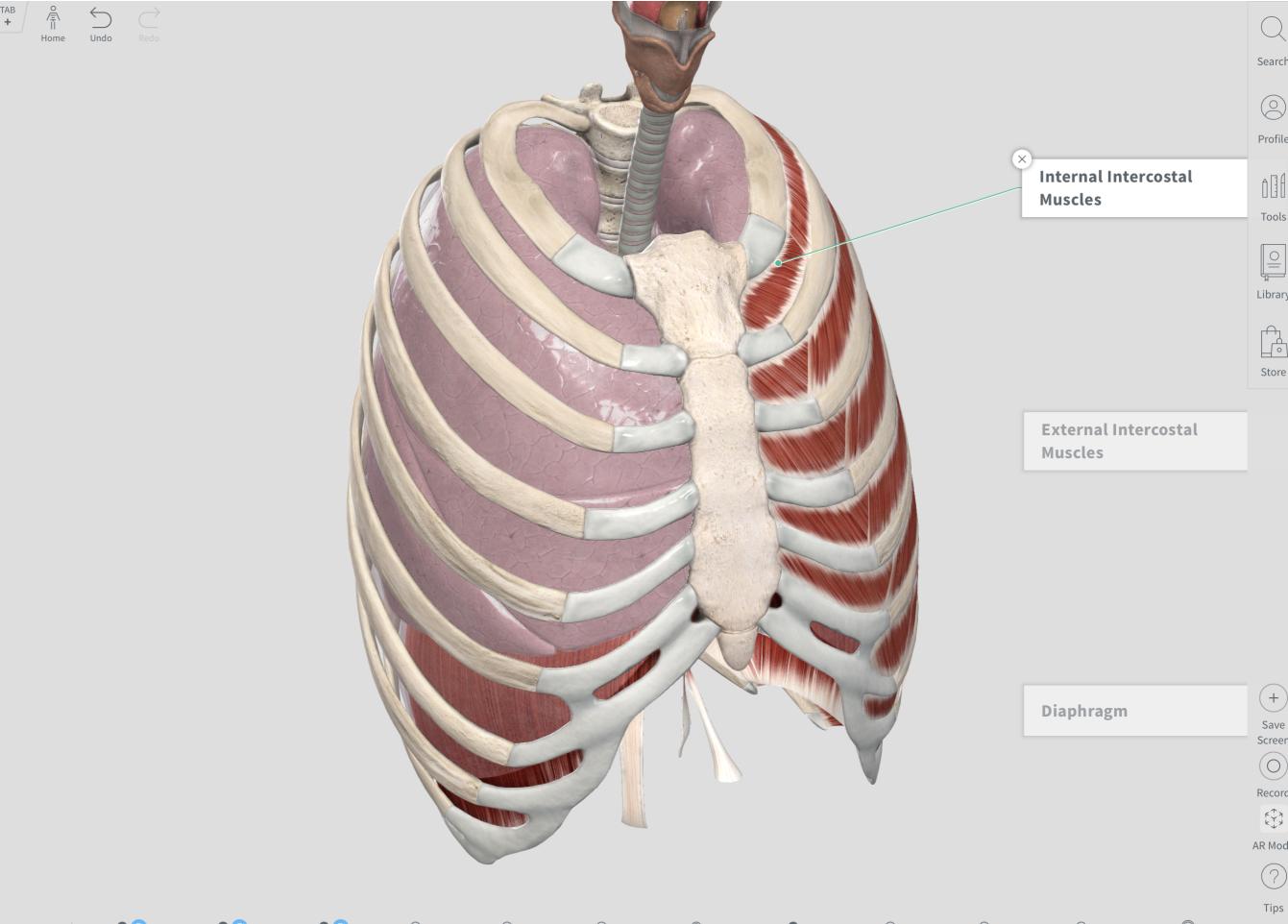
for the Pre-Hospital Provider



#### Course Objectives

- Identify respiratory anatomy
- Identify three gas laws that are involved in breathing
- Describe the physiological process of breathing
- Identify compliance and resistance of the airway
- Describe how to ventilate for compliance or resistance issues

#### Let's Look At The Chest Wall





























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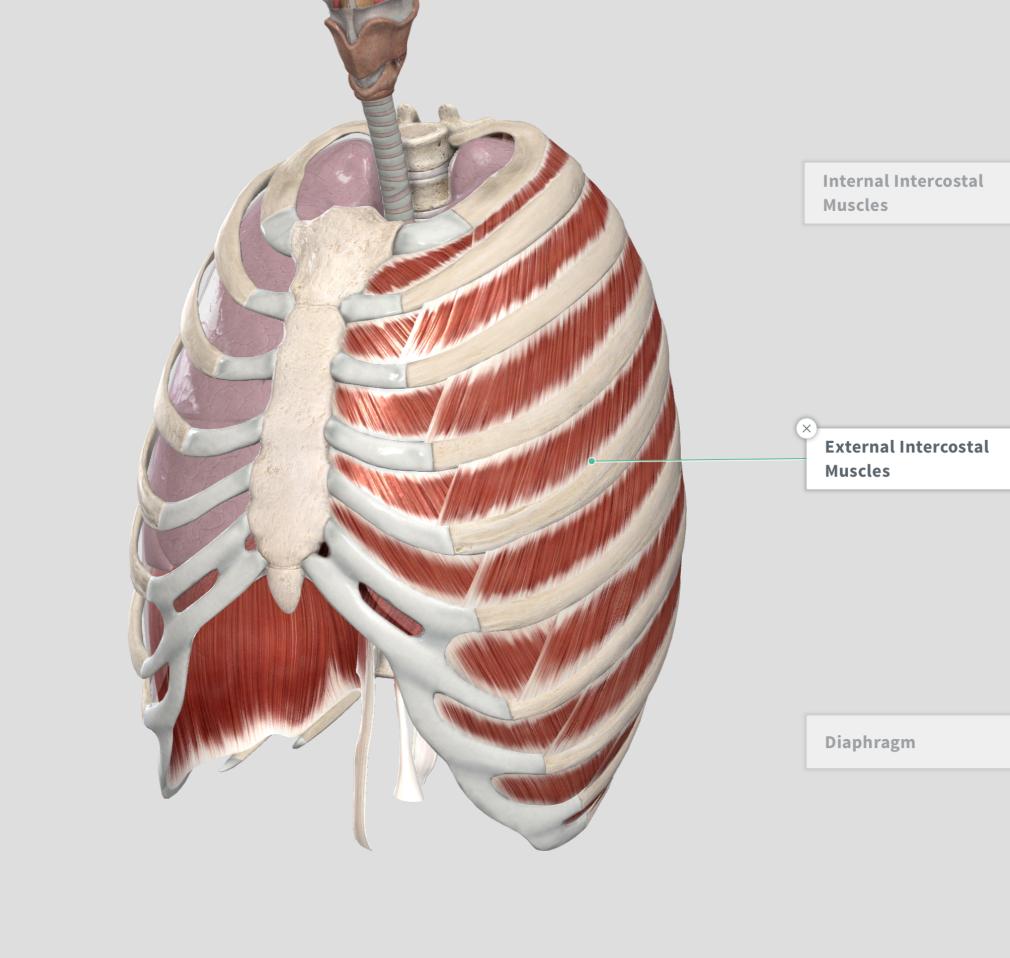








Tips





Home

























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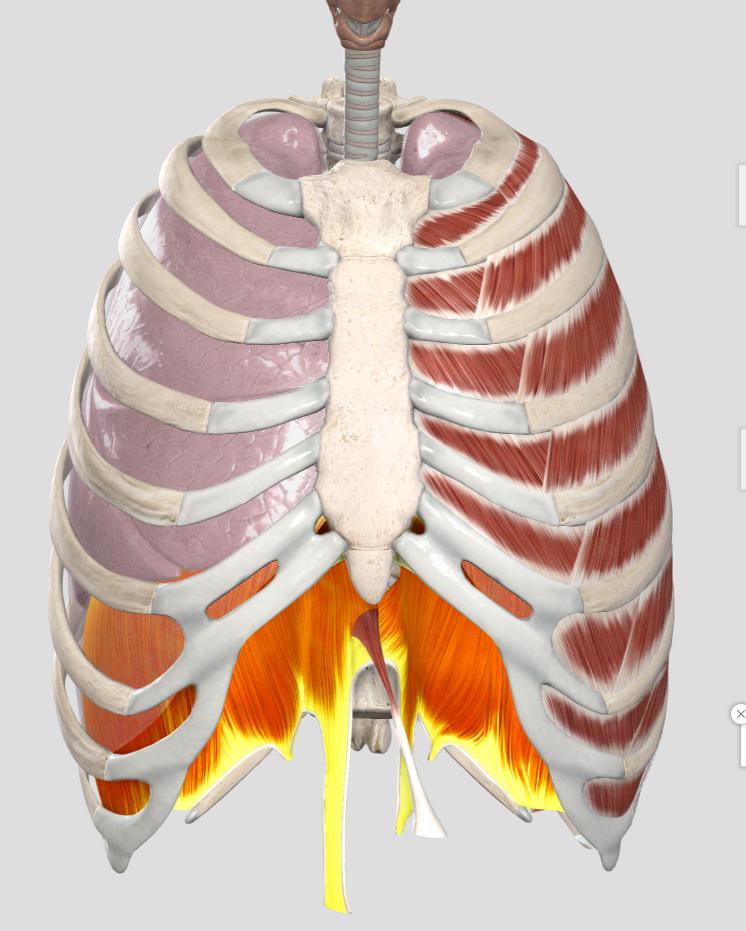
Save Screen

Record

 $\Leftrightarrow$ 

AR Mode

Tips



Internal Intercostal Muscles

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**External Intercostal** Muscles

Diaphragm



Save Screen







AR Mode



Tips

















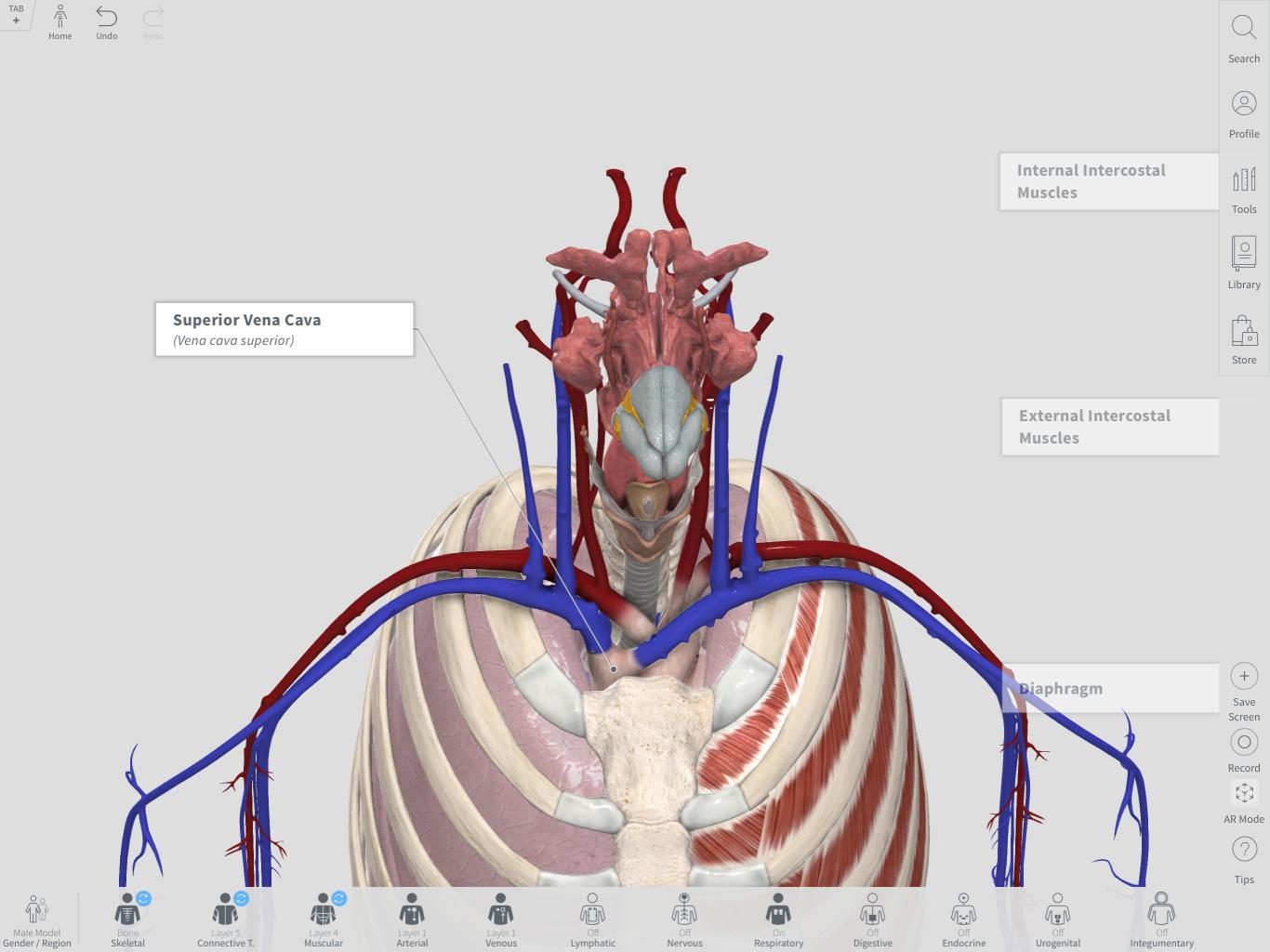




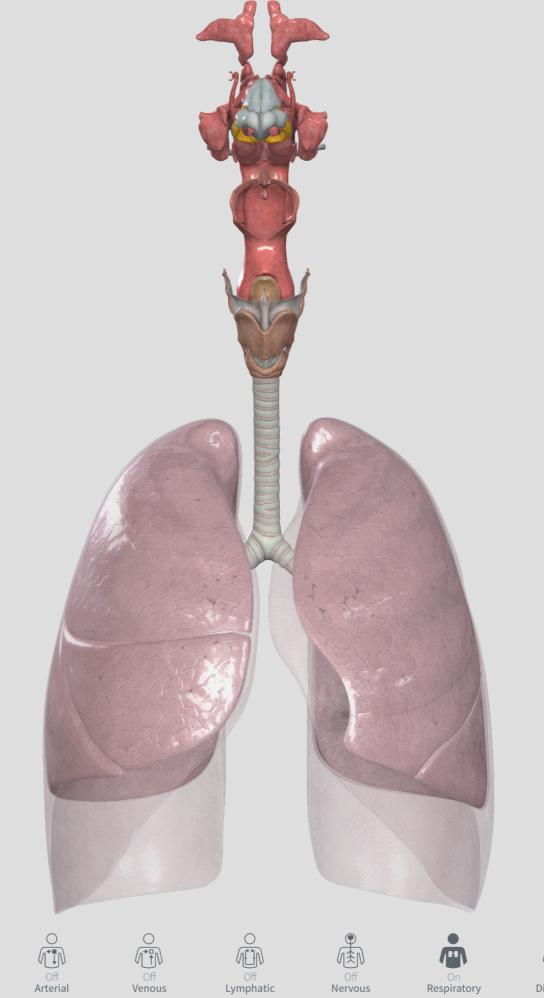








# Let's Look In The Chest Cavity



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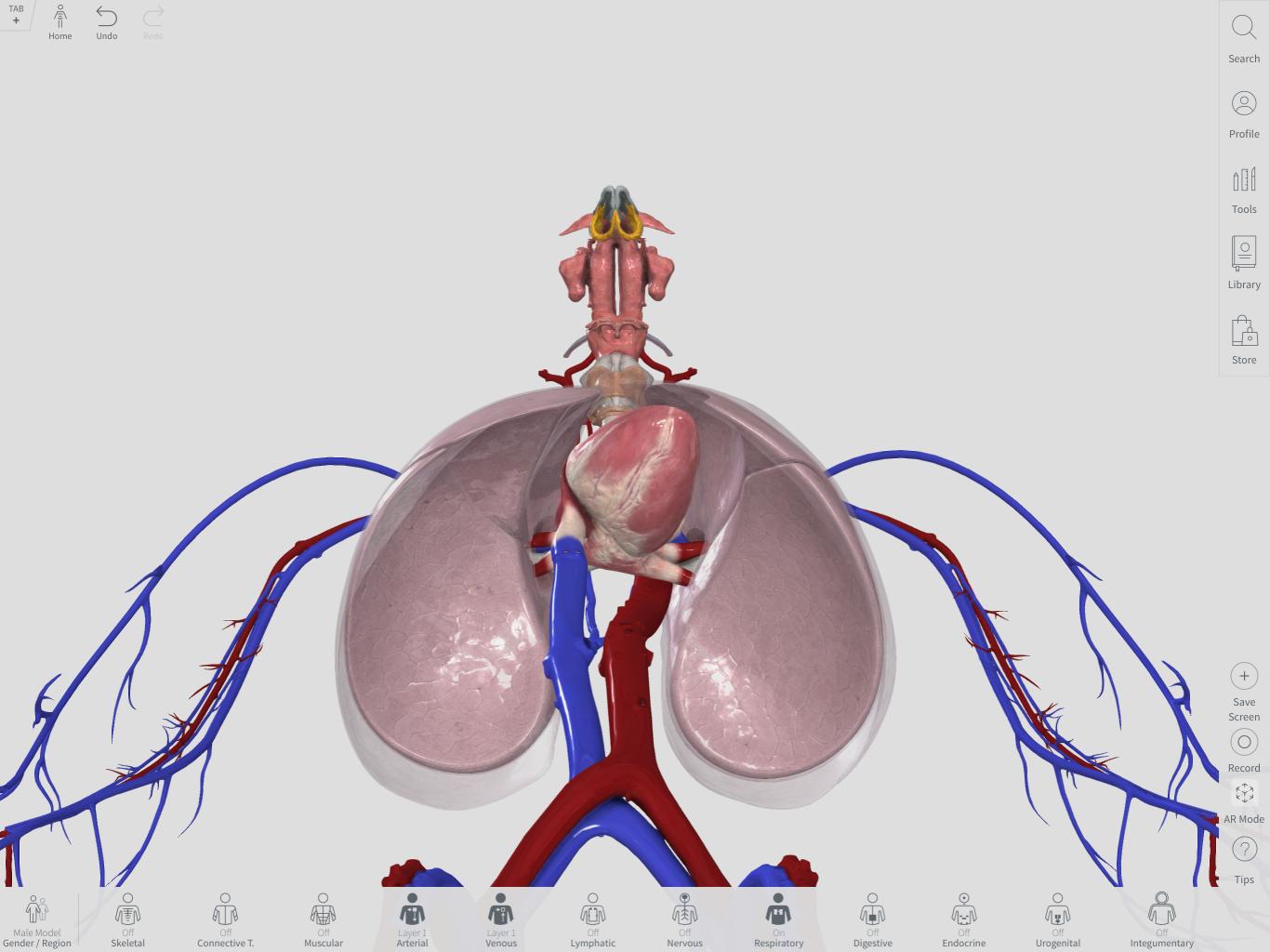


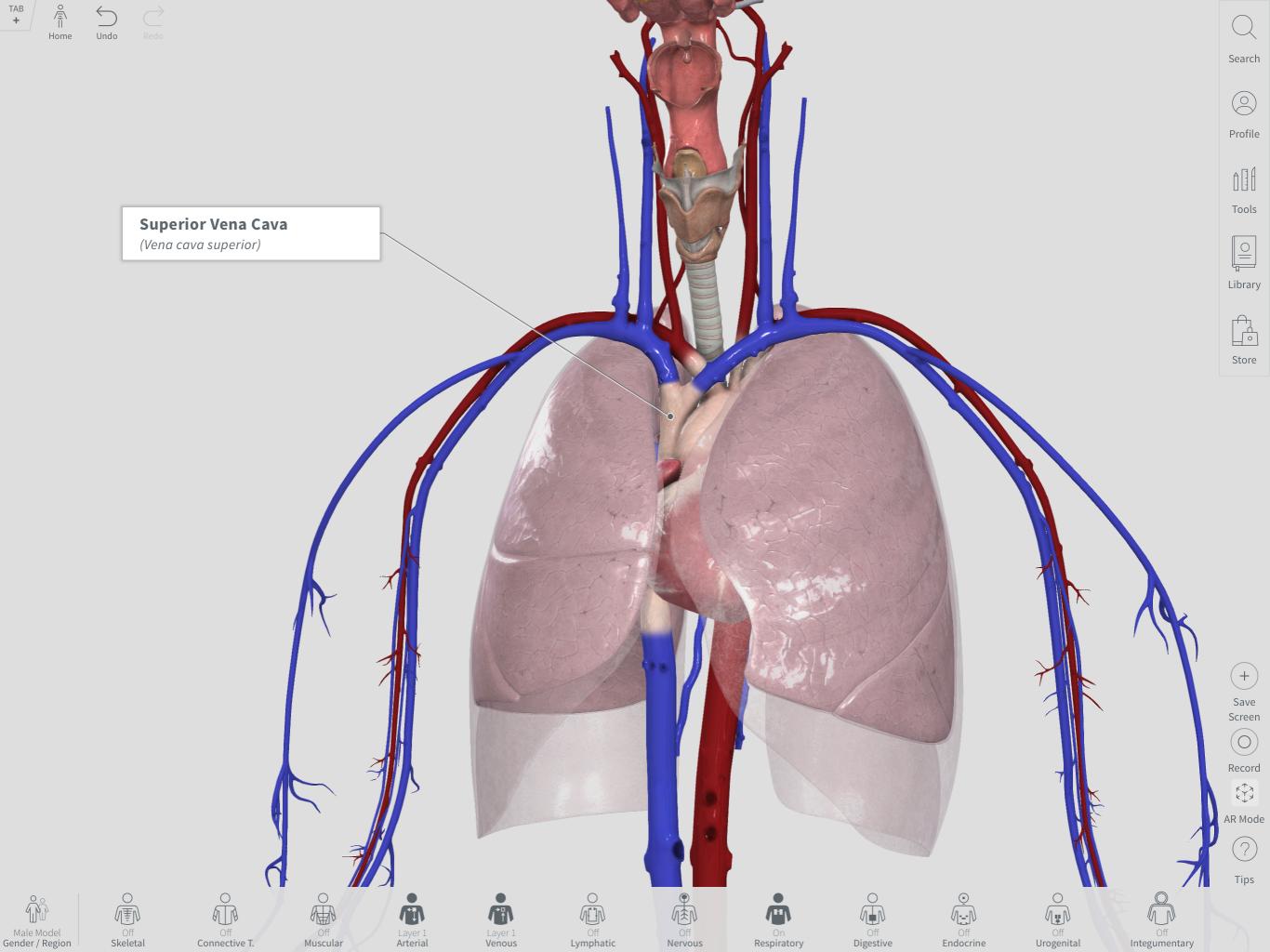


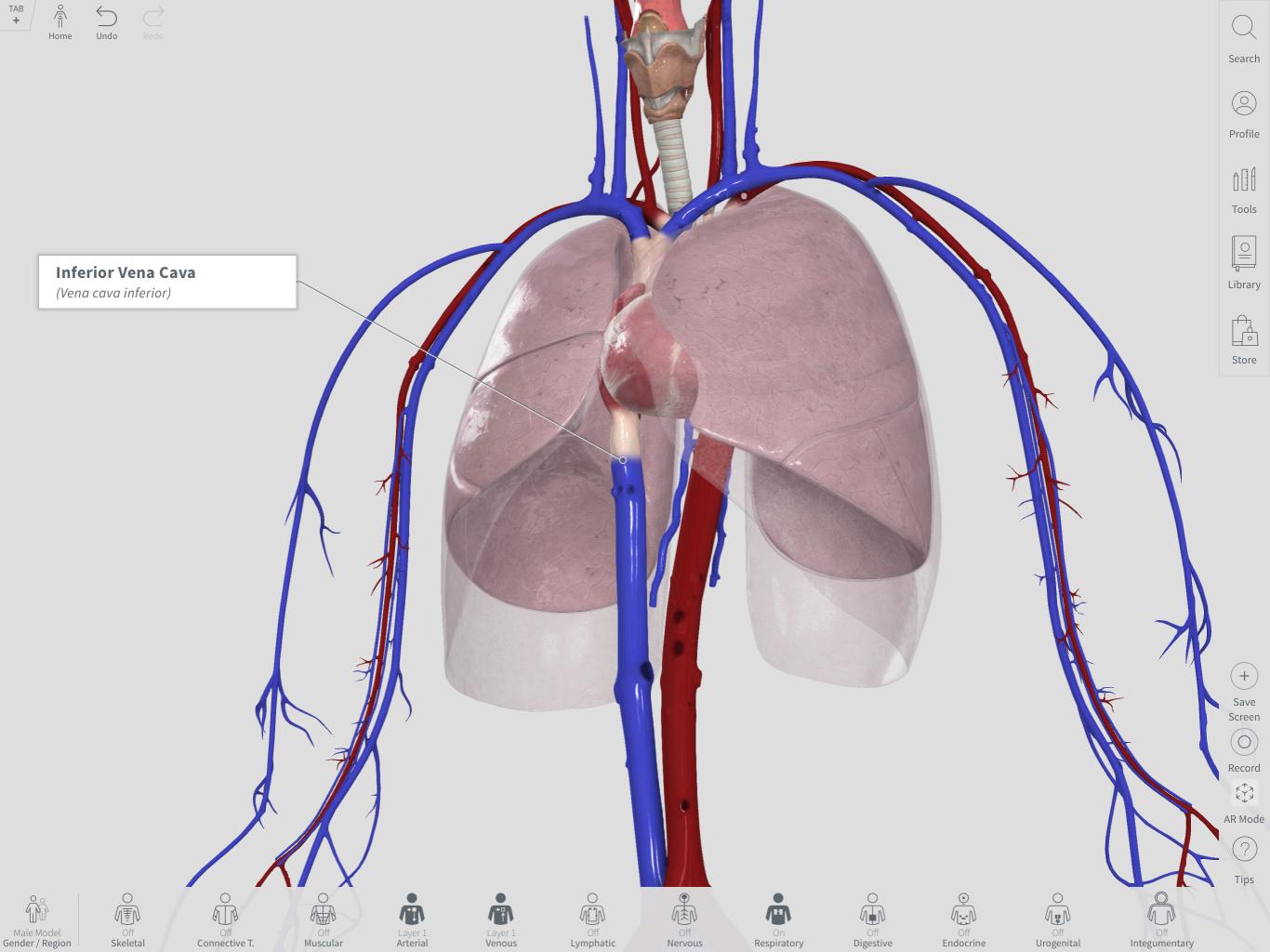




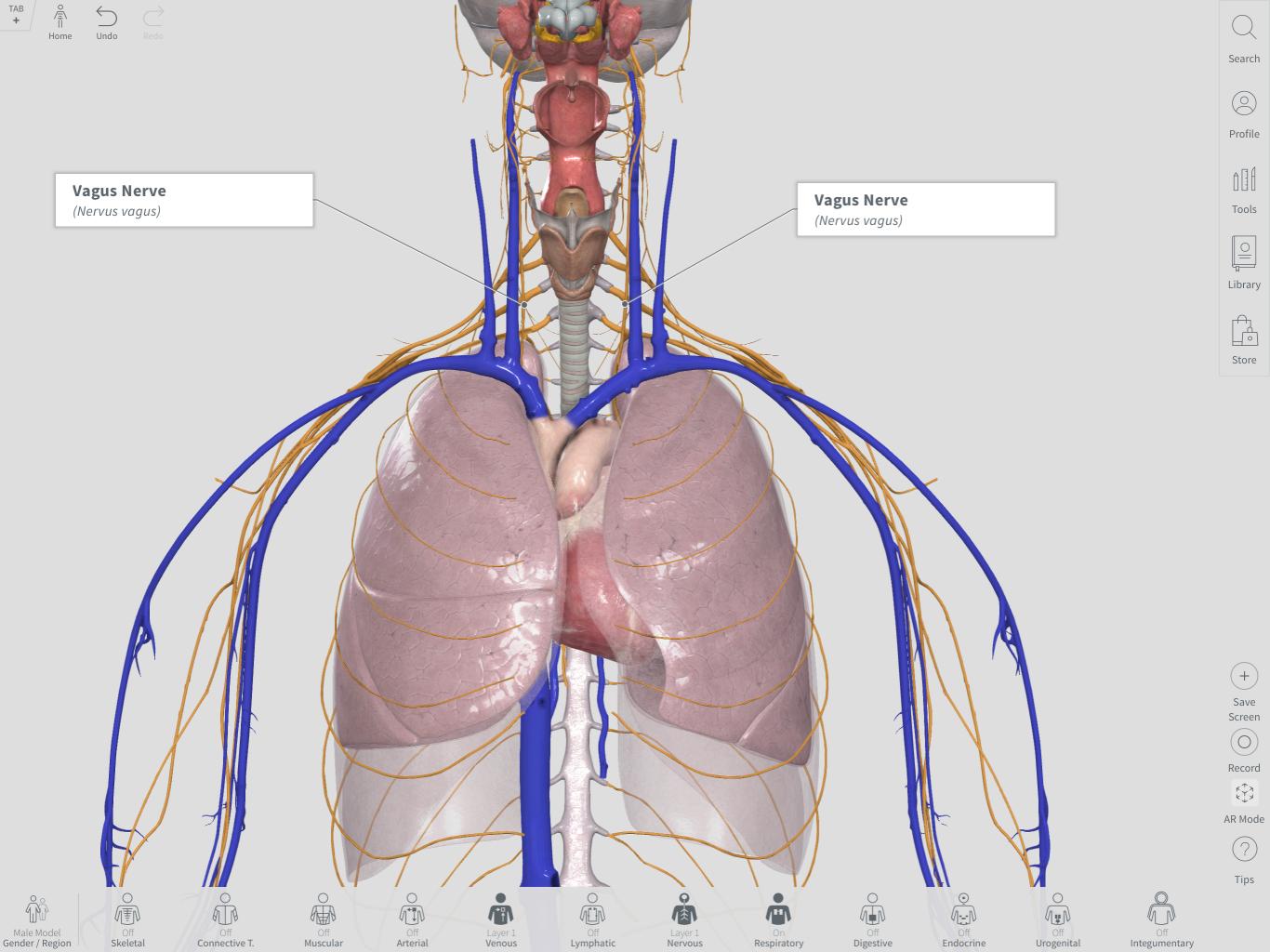


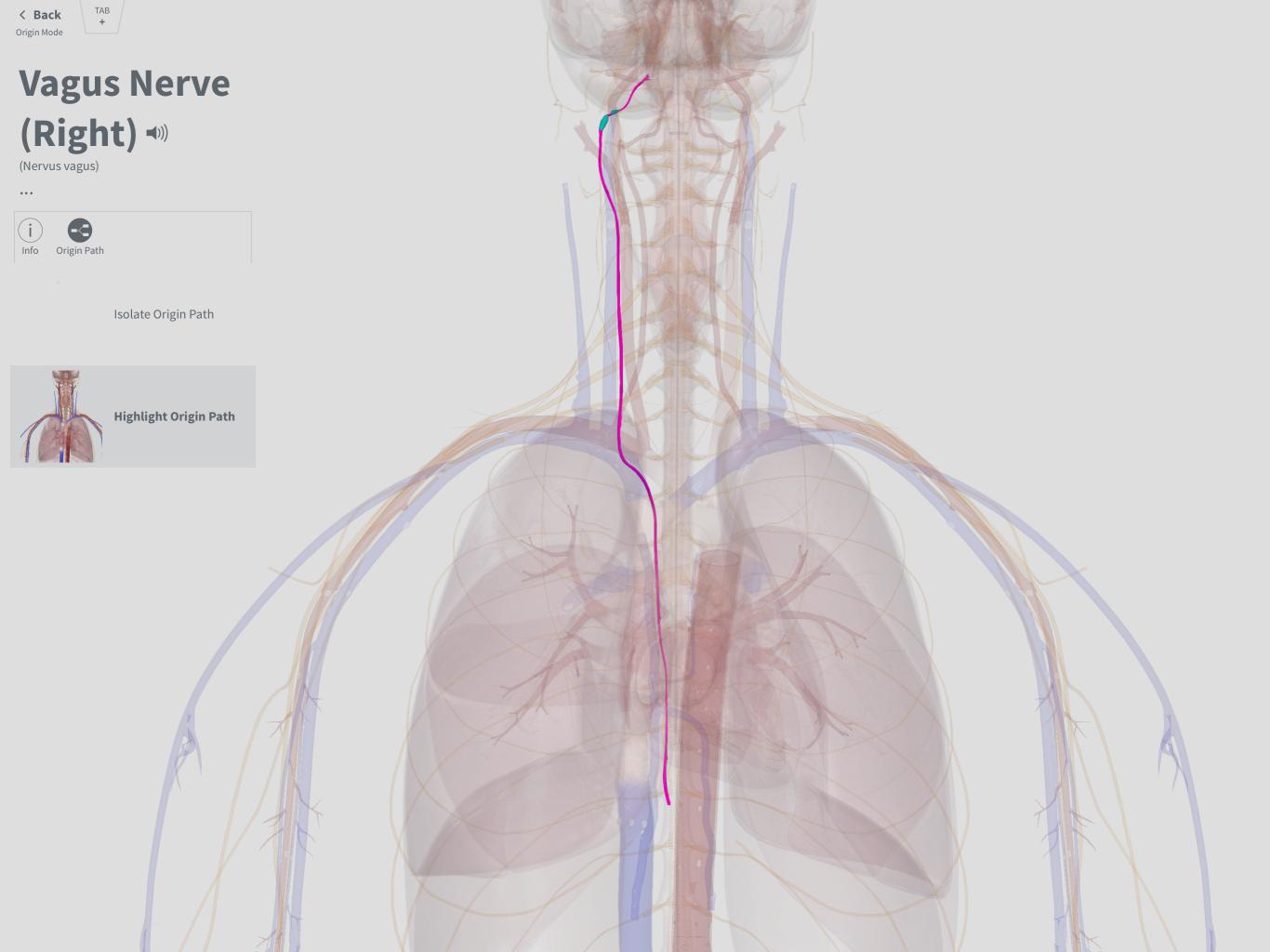




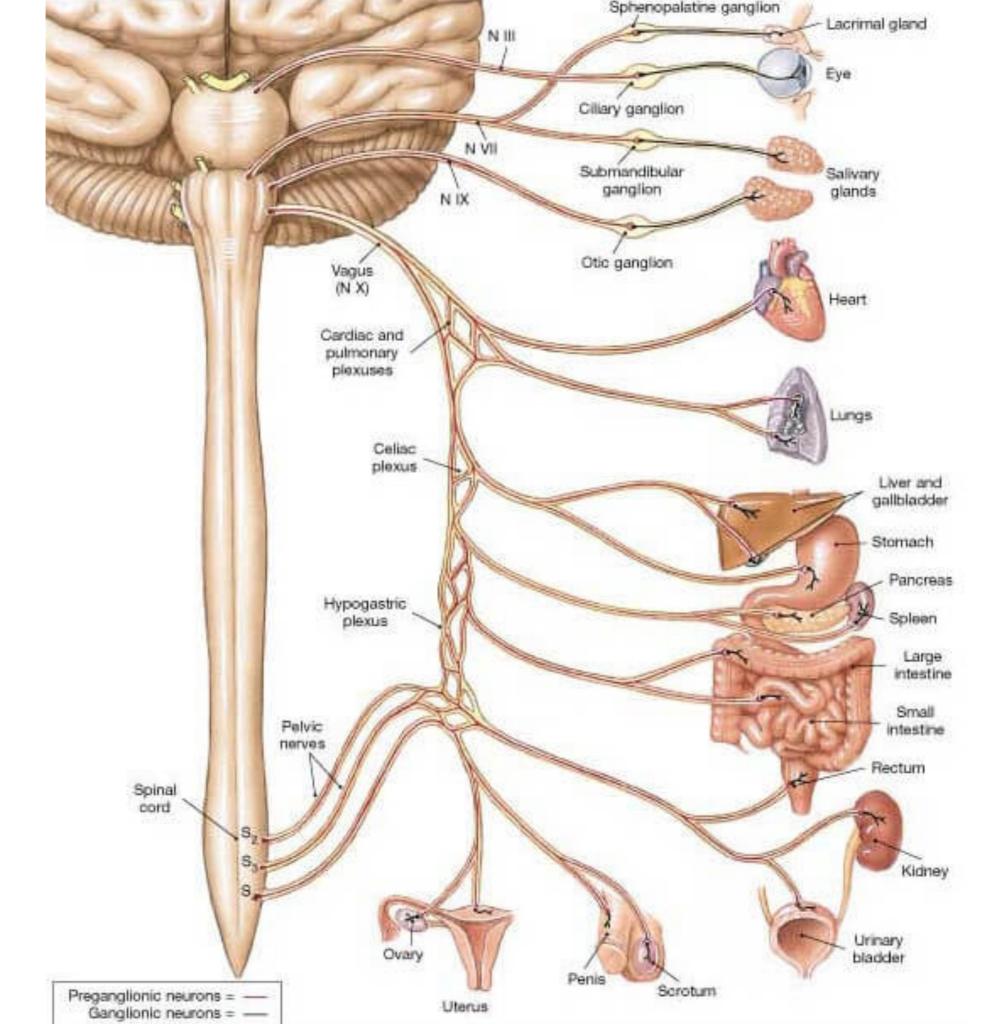


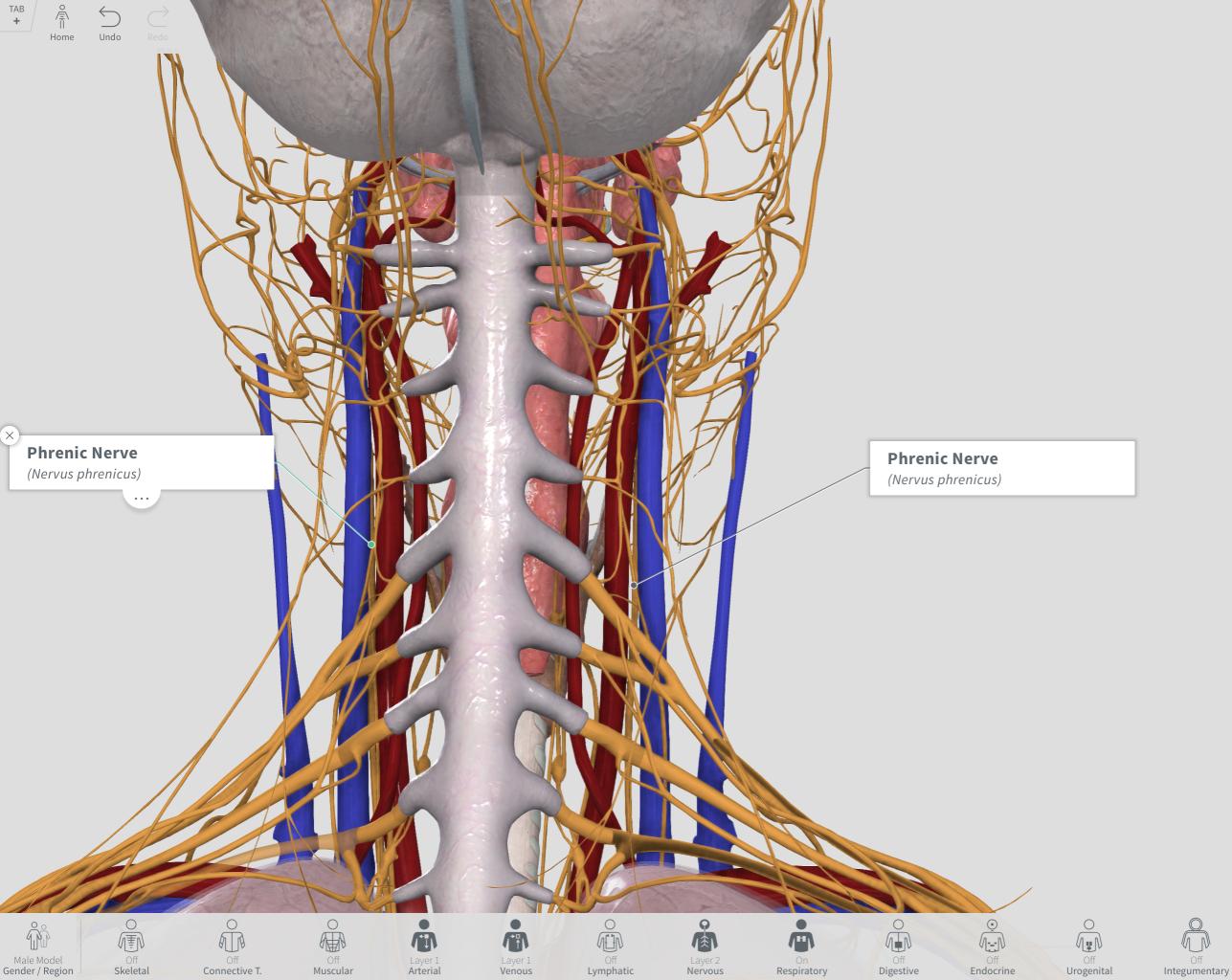
## Let's look At The Nerves

















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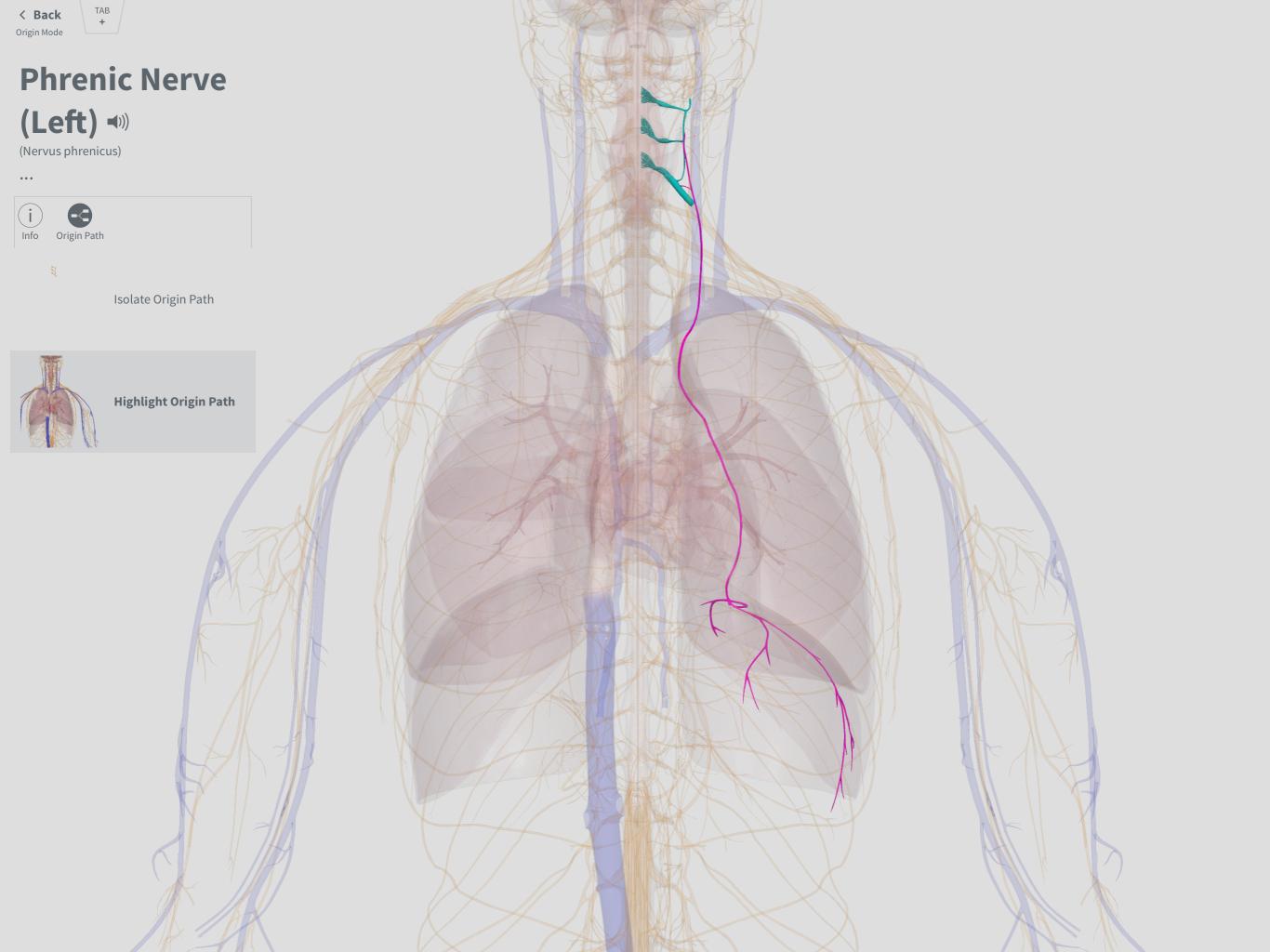


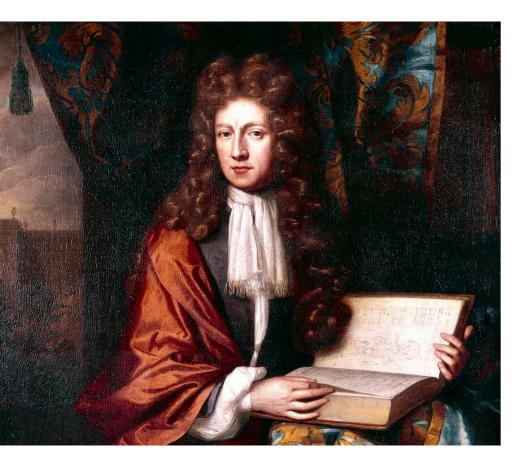
AR Mode

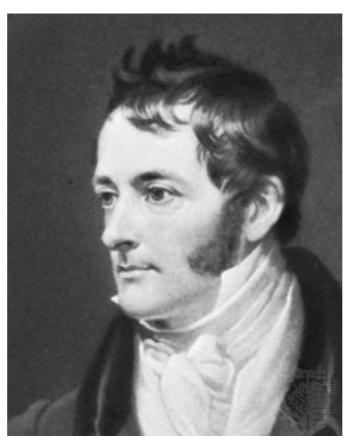


Tips









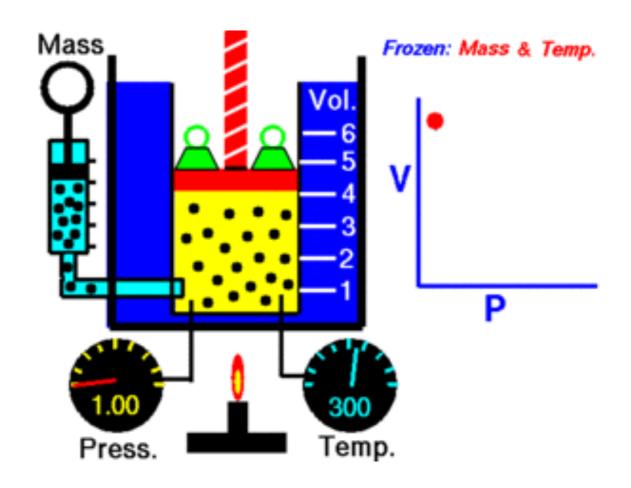


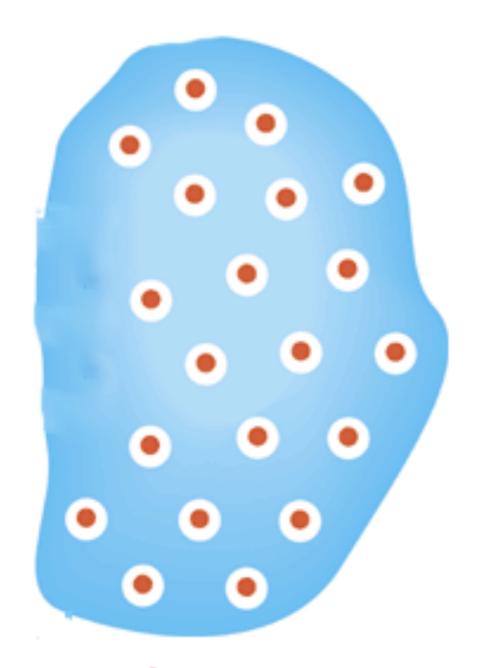
#### The Laws of Gases

Boyle's-Henry's-Charles's

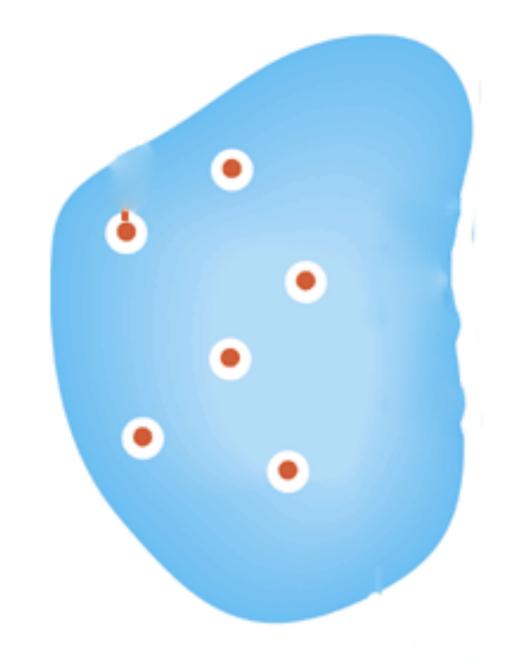
#### Robert Boyle's Law

- At a constant temperature
- Volume is inversely proportionate to pressure
- As pressure increases, volume decreases
- As pressure decreases, volume increases

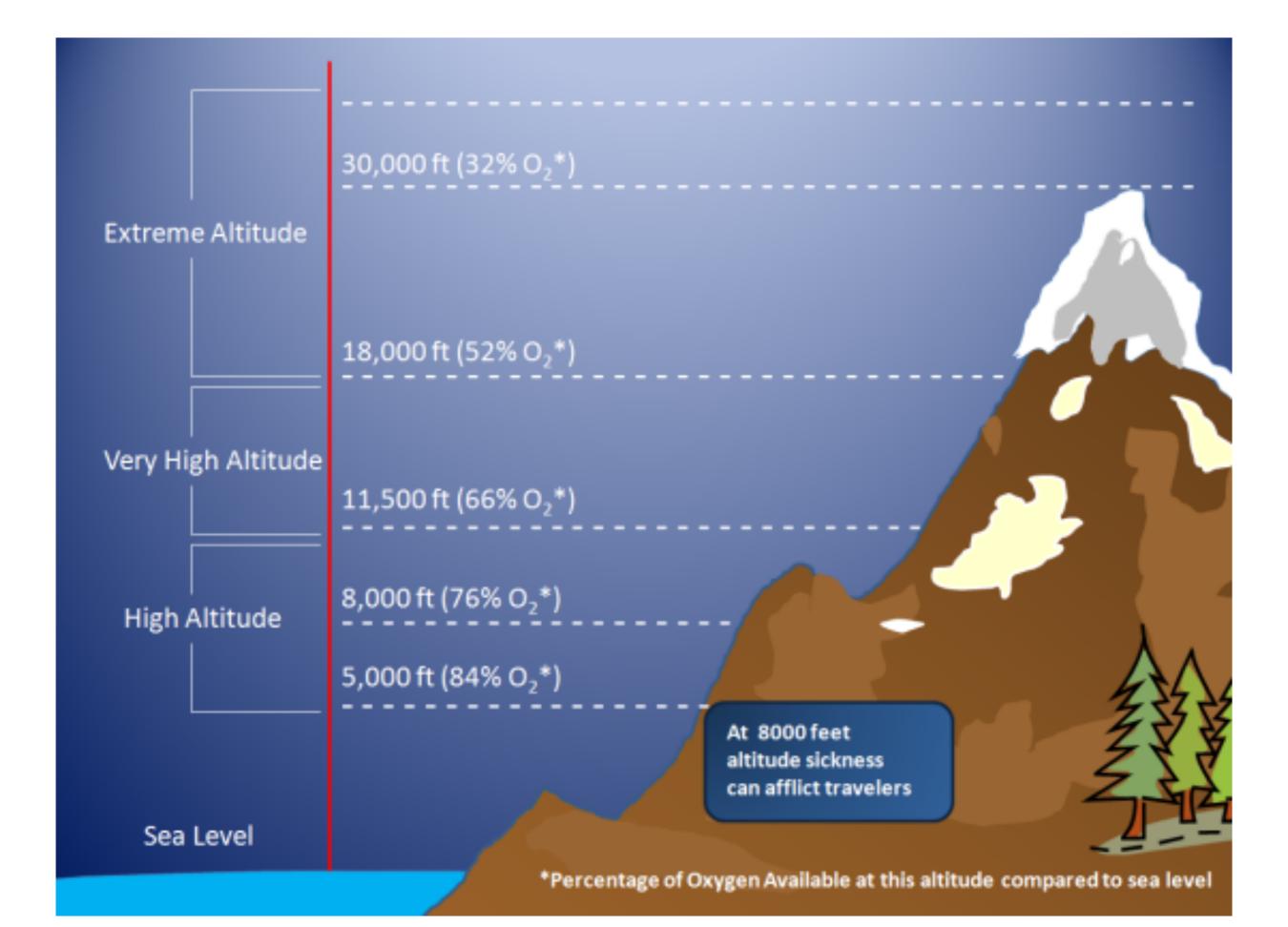




# of Oxygen in air at sea level

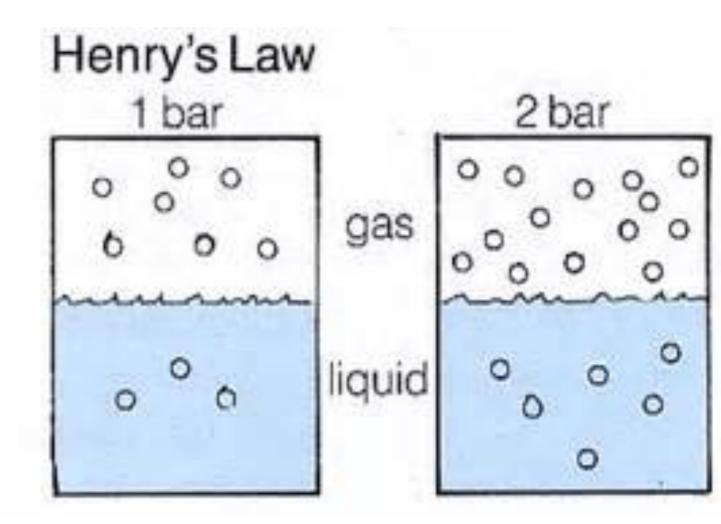


# of Oxygen in air at altitude

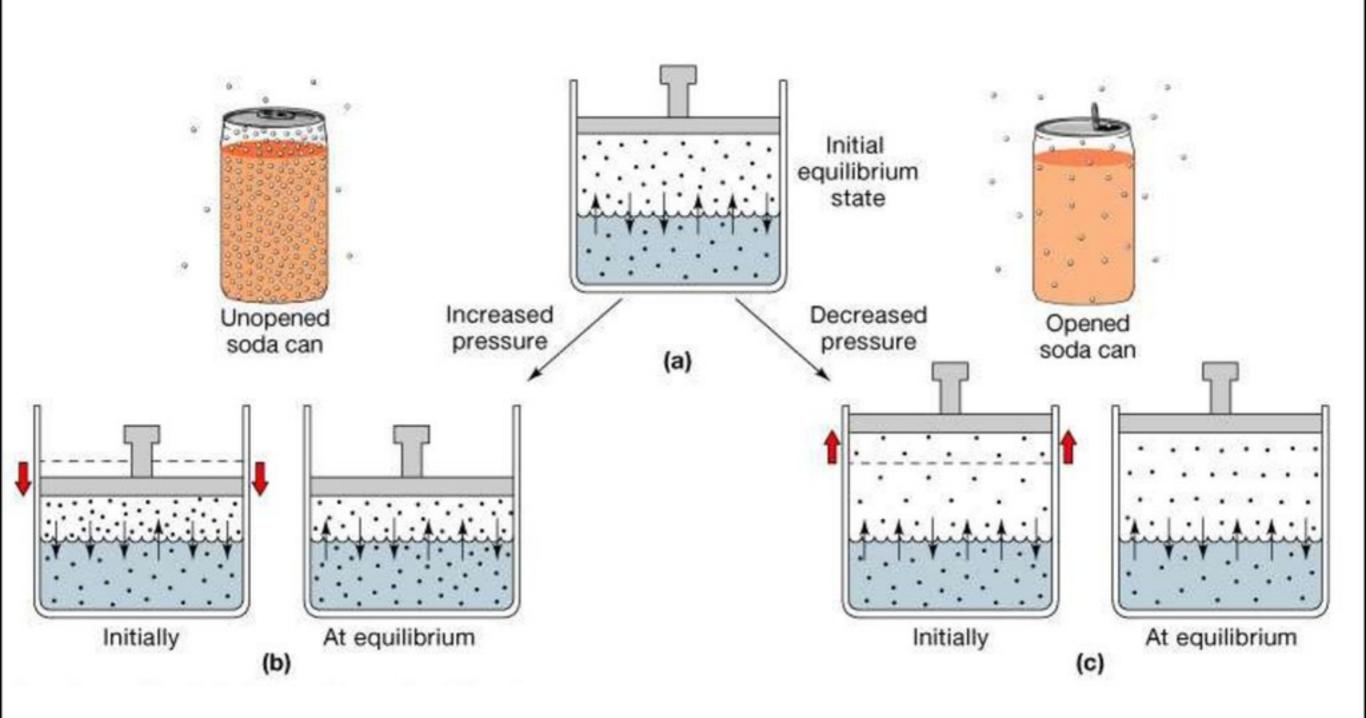


#### William Henry's Law

- The pressure of a gas above a liquid is proportional to the pressure dissolved in liquid at a constant temperature
- Huh? The more pressure applied. The more gas dissolves in liquid
- Think of a can of soda
- Before the can is open the air above the soda is almost pure CO2 until the can is opened

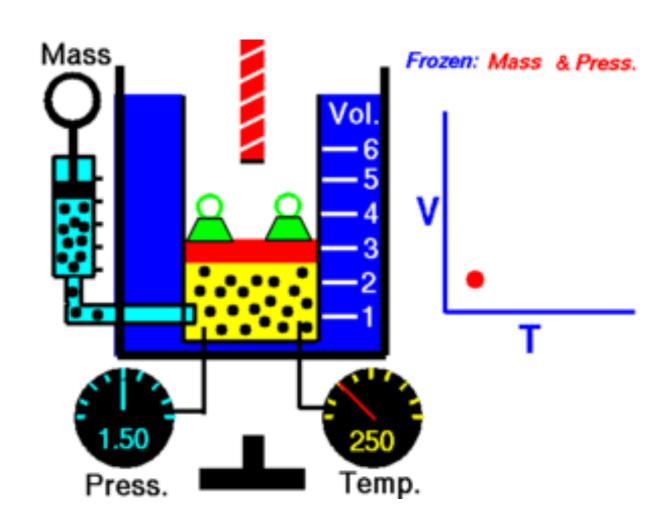


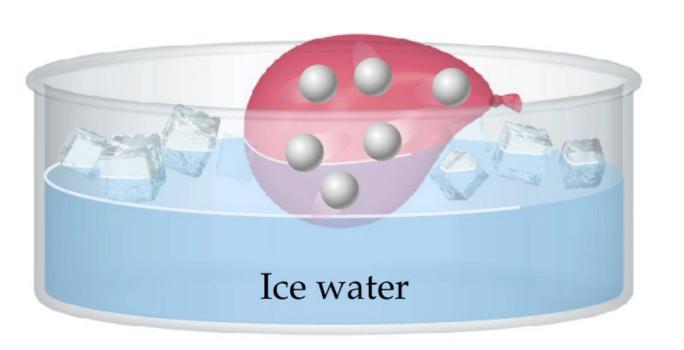
## Henry's Law and the Relationship between Solubility and Pressure

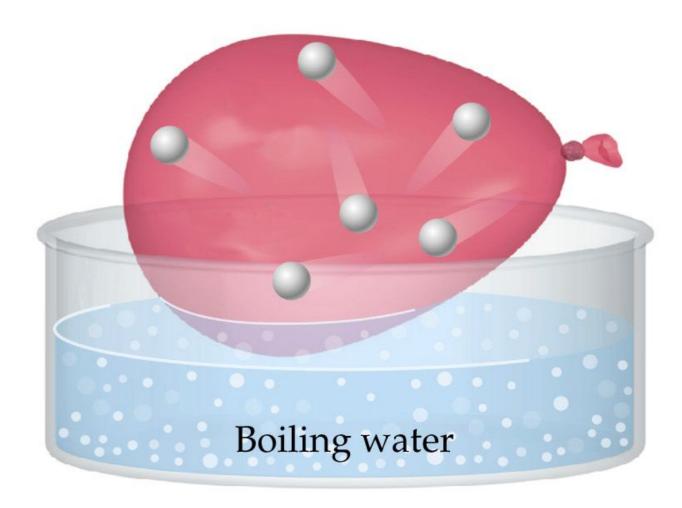


#### Jacques Charles's Law

- At a constant pressure, a gas volume increases as temperature increases
- Conversely, a gas volume decreases as temperature decreases



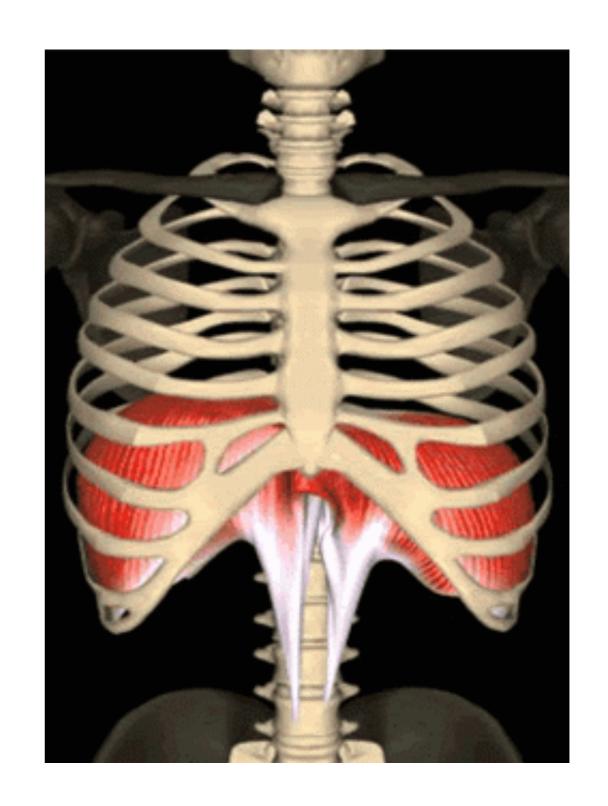




#### BREATHE

#### How We Breathe

- Diaphragm contracts
- Negative pressure created
- Normal breathing is 1-2 mmHg difference from thoracic to atmospheric pressure
- Creates normal sinus dysrhythmia

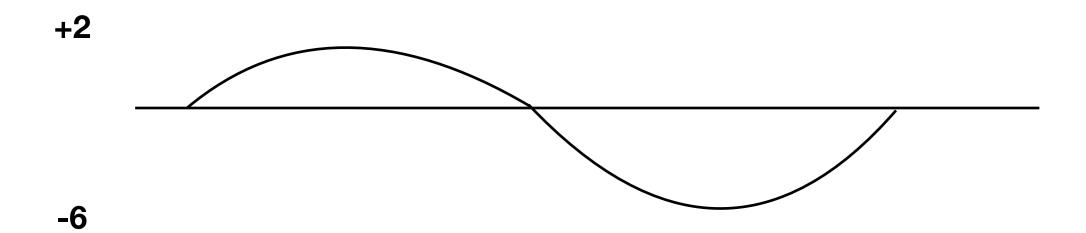




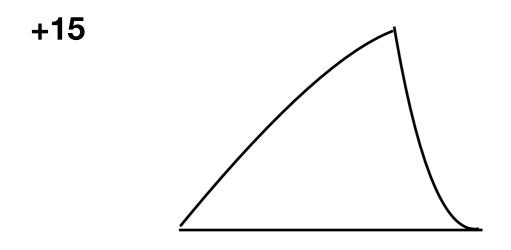
#### Positive Pressure

Ventilation

#### **Normal Ventilation (mm Hg)**



#### **Positive Pressure Ventilation (mm Hg)**



## Compliance & Resistance



### Compliance

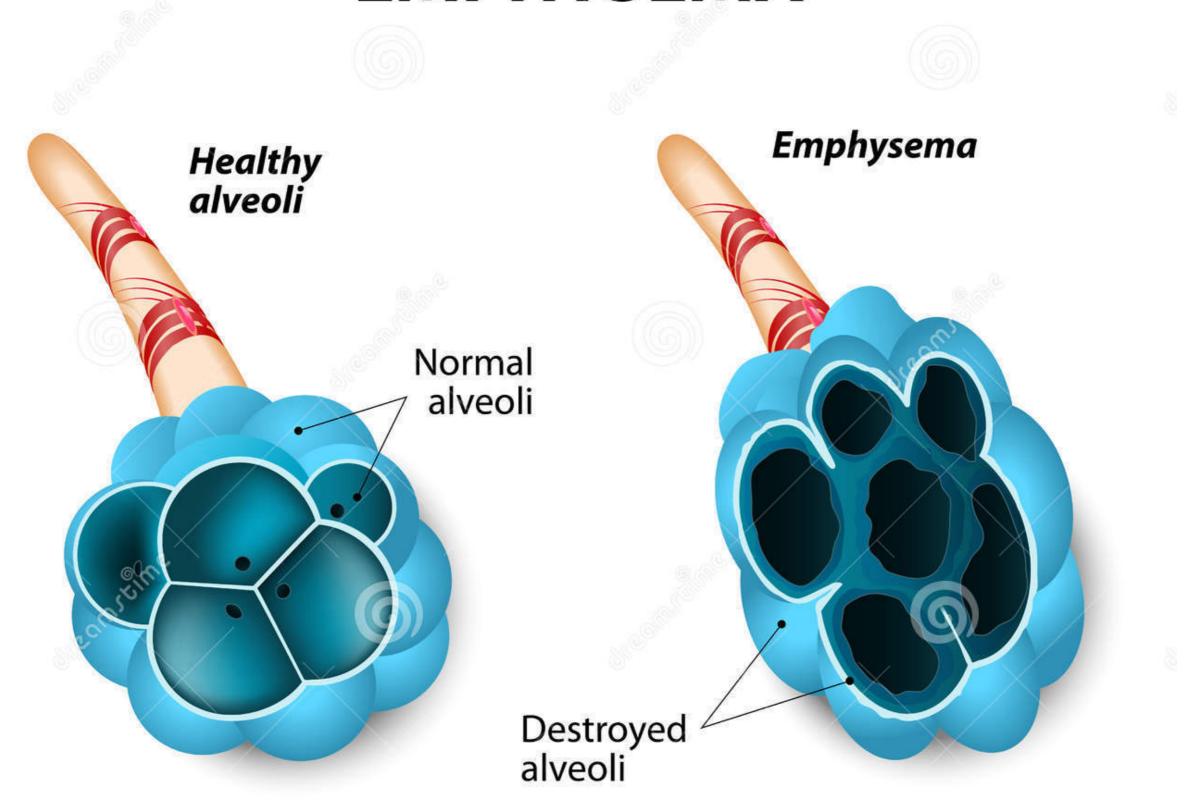
Problem with the alveoli

#### Causes of poor compliance

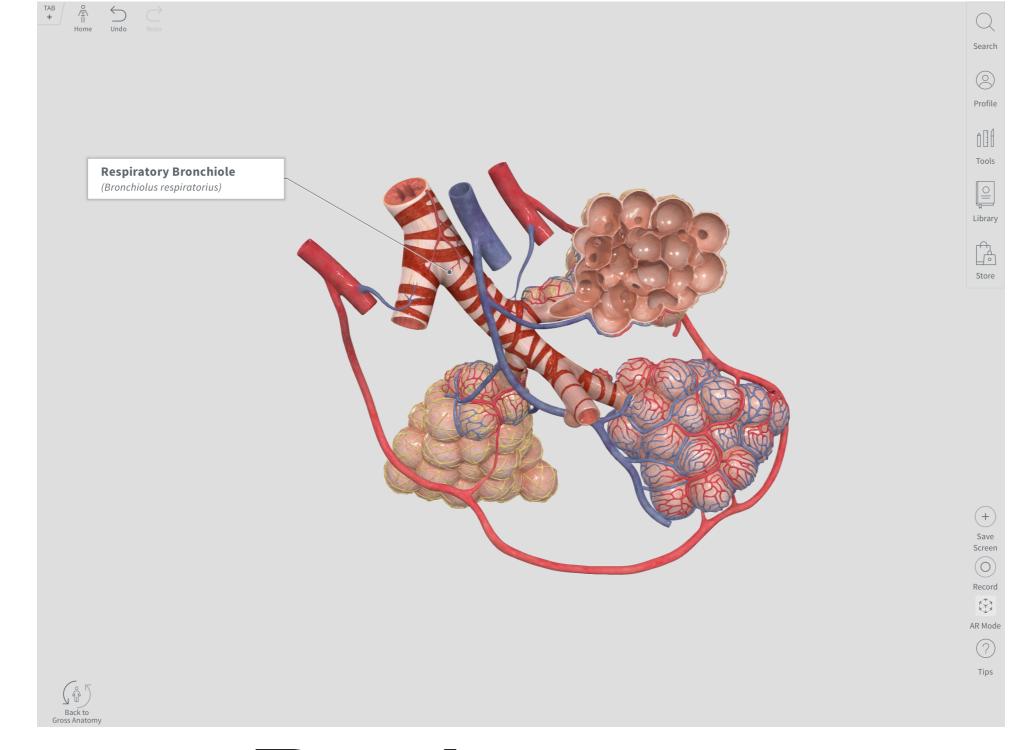
- Emphysema/COPD
- Pulmonary Edema
- Fibrosis
- Supine Position
- Any condition that looses elasticity

# Compliance issues can get air out but have a hard time getting air in

#### **EMPHYSEMA**







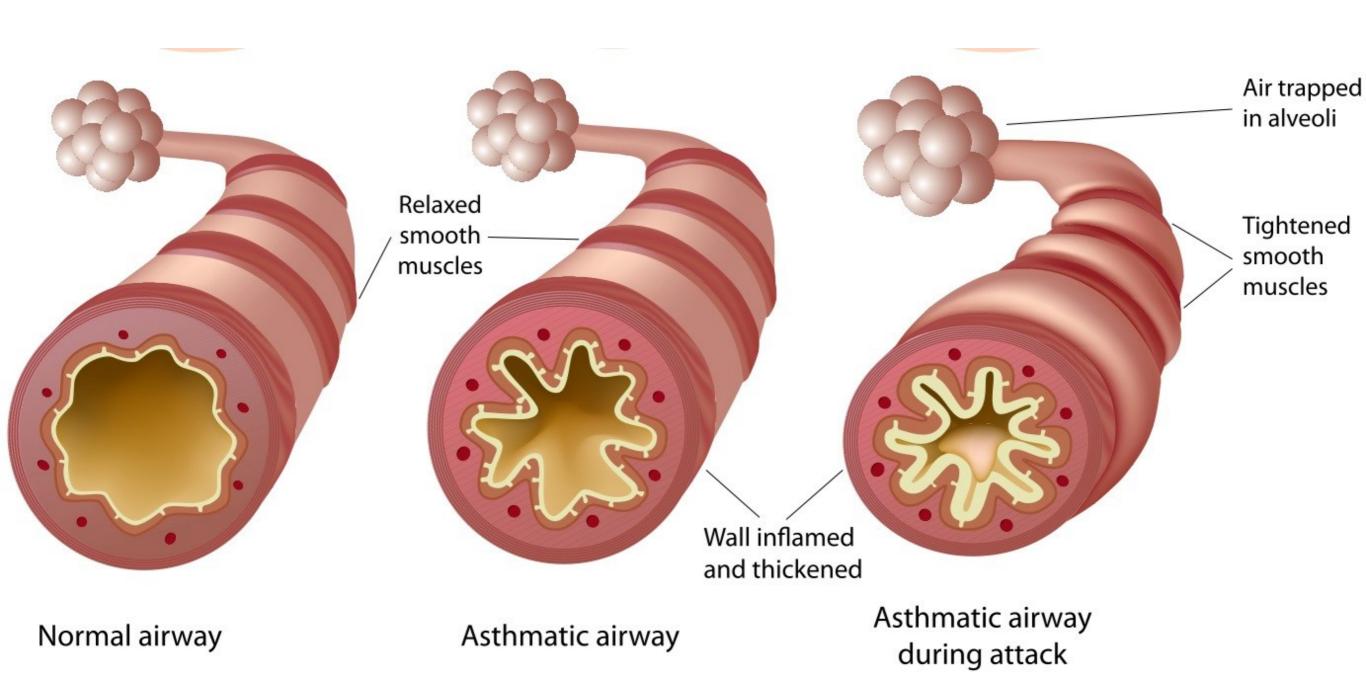
### Resistance

Problem with the tract

#### Causes of restriction

- Bronchoconstriction
- Emphysema on expiration
- Tracheal stenosis
- Mucous buildup

# Restrictive tracts can get air in but harder to get air out



#### Ventilation



-Peter Sloterdijk

#### Important Reminders

- Respiratory Rate (RR) & Tidal Volume (Tv)= Controls CO2
- I:E Ratios help control pressure with BVM
- Keep Peak Inspiratory Pressures Below 40 mm Hg
- Use Manometer or Pressure Relief Valve





## Ventilation for Poor Compliance

- Increase inspiratory times with shorter to normal expiratory times.
- Example: normal I:E 1:2, Compliance 1-4:1-2
- Poor Compliance will feel easy to bag at first then become increasingly harder and Peak Inspiratory Pressures (PIP) will tend to increase gradually.

### Ventilation for Restriction of the Tract

- Inverse I:E ratios. Quick inspiratory long expiratory phase
- Example: normal I:E 1:2, Restrictive Tract 1 or less:2-5
- Restrictive Tract will feel difficult to ventilate from initial squeeze of the bag and have high initial Peak Inspiratory Pressures (PIP)

#### Questions?